

Extraction of pyrethrins from Dalmatian pyrethrum (*Tanacetum cinerariifolium* /Trevir./ Sch. Bip.) by matrix-solid phase dispersion

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Fig. 1. Dalmatian pyrethrum (*Tanacetum cinerariifolium* /Trevir./ Sch. Bip.)

Dalmatian pyrethrum (*Tanacetum cinerariifolium* /Trevir./Sch. Bip.) is a plant species of the Asteraceae family, restricted to the eastern coast of the Adriatic Sea. It is a perennial plant species known as the source of natural insecticide pyrethrin. The term pyrethrin refers to the six insecticide active components: pyrethrin I and II, cinerin I and II, jasmolin I and II. Insecticidal potential of Dalmatian pyrethrum is measured by the content and composition of pyrethrins which are mainly concentrated in the flower heads.

In order to obtain deeper knowledge of its insecticidal potential, content and composition of pyrethrins was assessed by matrix-solid phase dispersion (MSPD). MSPD, a process allowing simultaneous extraction and clean-up of analytes from solid samples, is used as an alternative technique to classical extraction methods with significant reduction in solvent consumption. The performance of MSPD is based on blending analyzed samples (solid or semi-solid) with a suitable adsorbent to form a homogenous packing material. After successful packing, the sample/adsorbent column is eluted by a stepwise solvent program.

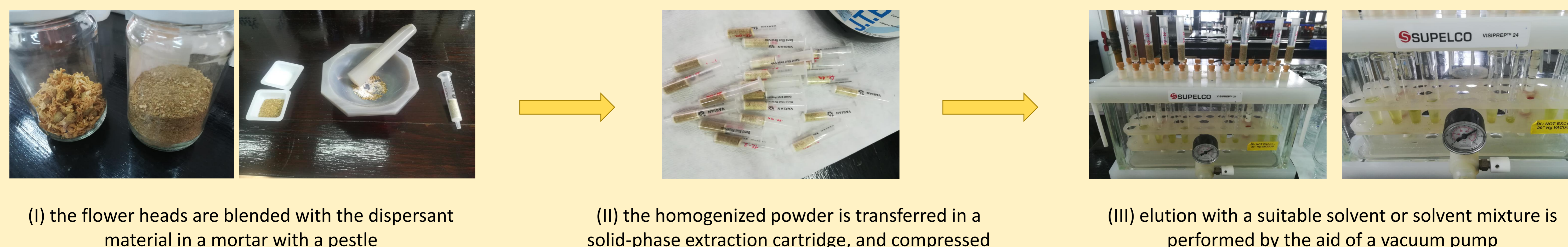


Fig. 2. Main steps of matrix-solid phase dispersion extraction procedure

1st set of experiments:

- (I) flower heads:florisil = 1:2
flower heads:C18 = 1:2
flower heads:silikagel = 1:2
- (III) 10 mL of ethanol
10 mL of acetone
10 mL of acetonitrile
10 mL of ethyl acetate
10 mL of dichloromethane

2nd set of experiments:

- (I) flower heads:florisil = 1:2
flower heads:silikagel = 1:2
- (III) 10 mL of acetone:ethyl-acetate = 1:1 (v/v)
10 mL of acetone:ethyl-acetate = 1:2 (v/v)
10 mL of acetone:ethyl-acetate = 1:4 (v/v)
10 mL of acetone:ethyl-acetate = 2:1 (v/v)
10 mL of acetone:ethyl-acetate = 4:1 (v/v)

Table 1. Comparison of recoveries obtained using different sorbents and solvents ($n = 3$) in the first set of experiments

pyrethrins	Recovery (%)											
	FLORISIL				SILIKAGEL				C18			
	ethanol	acetone	acetonitrile	ethyl-acetate	ethanol	acetone	acetonitrile	ethyl-acetate	ethanol	acetone	acetonitrile	ethyl-acetate
cinerin II	40.77	76.41	80.61	83.26	21.93	92.59	63.68	109.14	24.42	33.69	70.16	81.91
pyrethrin II	53.60	83.39	66.96	79.54	49.32	91.47	66.94	106.69	57.39	73.85	70.08	73.26
jasmolin II	58.95	88.07	93.32	80.75	34.29	85.30	58.26	58.19	58.79	70.22	77.74	81.90
cinerin I	48.49	91.28	82.94	88.91	22.59	83.95	60.08	108.30	20.33	26.48	23.01	30.11
pyrethrin I	50.89	83.09	73.19	85.46	52.04	84.44	68.35	99.79	51.14	62.49	59.11	67.22
jasmolin I	50.44	91.05	88.12	75.90	45.16	95.35	57.54	102.38	58.16	69.88	38.75	78.93

*Recoveries obtained using dichloromethane as elution solvent were very low for all six pyrethrins (data not shown)

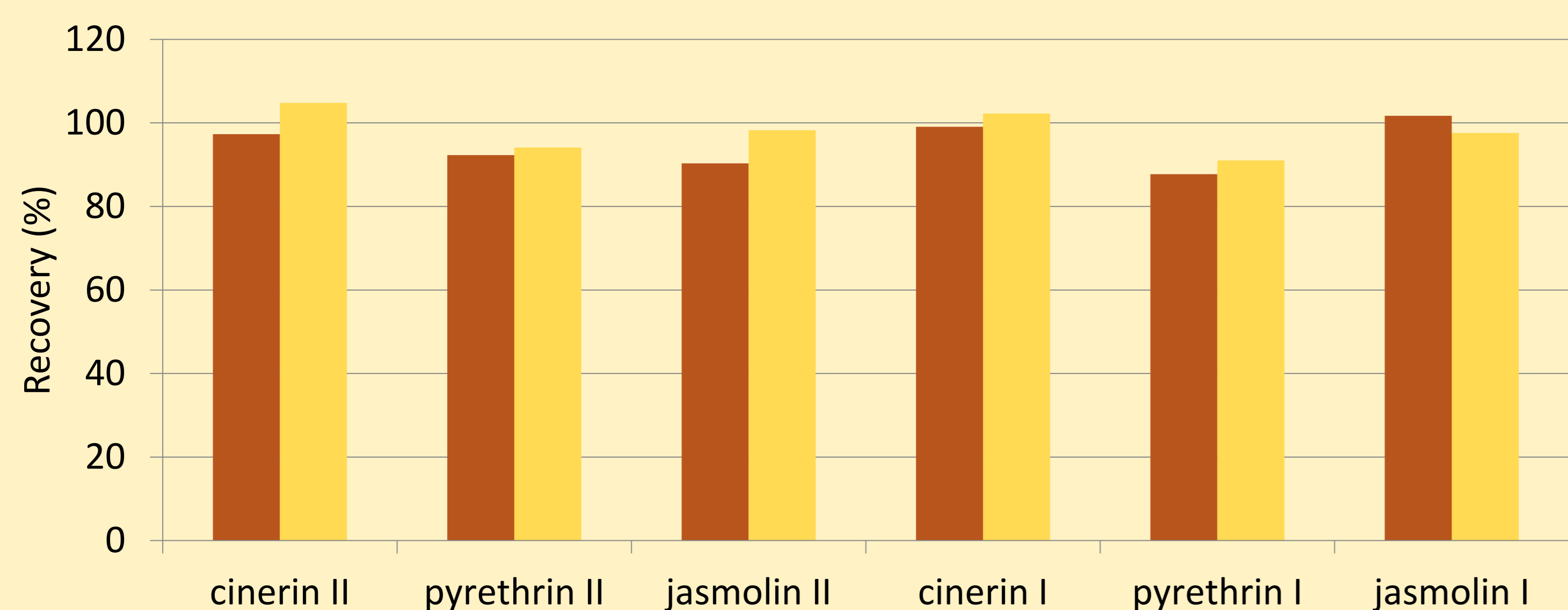


Fig. 3. Recoveries obtained using mixture of acetone:ethyl-acetate = 1:1 (v/v) and florisil (lighter column) or silikagel (darker column) sorbent ($n = 3$) in the second set of experiments

All six pyrethrins were isolated from flower heads of Dalmatian pyrethrum using MSPD followed by HPLC-DAD. Results showed that pyrethrins I and pyrethrins II were successfully extracted by florisil as a sorbent using 10 mL of acetone:ethyl-acetate = 1:1 (v/v) as an elution solvent using 1:2 ratio of flower heads and sorbent. Presented results prove that MSPD can be attractive, affordable and effective method for the extraction of pyrethrins from Dalmatian pyrethrum.

Conclusion